

Using Isotope Hydrology Tools to Support Conjunctive Water Management

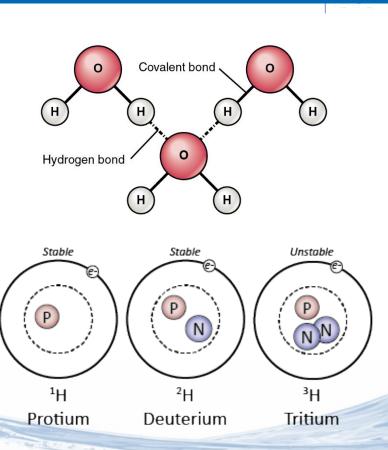
Jodie Miller

Section Head, Isotope Hydrology, Department of Nuclear Sciences and Applications, IAEA



The Water Molecule

- Water is a polar inorganic compound that is at room temperature a tasteless and odorless liquid, nearly colorless with a hint of blue.
- Density: 997 kg/m³
- Boiling point: 212°F (100°C)
- Molar mass: 18.01528 g/mol.
- Melting point: 32°F (0°C)
- What happens when we consider water in terms of an isotopic molecule?





Tools We Can Use: Isotope Hydrology



Progeny

Isotope

Decay

Process

Stable Isotopes

- Abundance constant, but distribution changes
- Used to quantify interactions between different reservoirs

Radioactive Isotopes

 C_{t0}

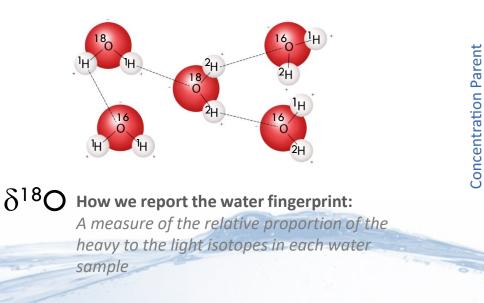
 C_{t1}

Time

- Concentration affected by radioactive decay
- Used to examine the duration of processes

Parer

Isotope

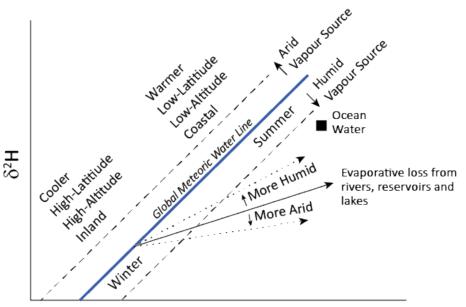




Meteoric Water Line is a way to relate the isotopes of the hydrogen and oxygen in the water molecule in precipitation.

Precipitation is the source input to surface water and groundwater systems

The precipitation $\delta^{18}O$ and $\delta^{2}H$ signal is particularly sensitive to fluctuations in temperature



 δ^{18} O

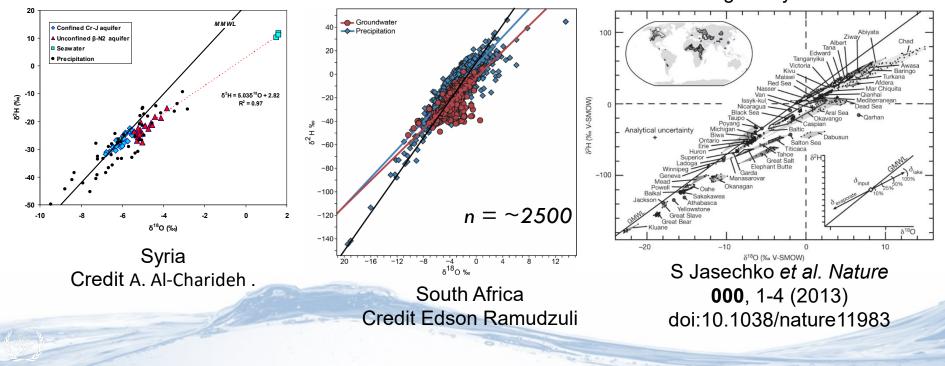
Sampling Strategies and Data Availability



 δ^{18} O and δ^{2} H values of groundwater and precipitation.

 $\delta^{18}O$ and $\delta^{2}H$ values of precipitation vs groundwater.

 δ^{18} O and δ^{2} H values of large lakes and semi-enclosed seas globally.



Diversity of Water Isotope Fingerprinting





Making Decisions on Available Data and Information



Uncertain Uncertain Dry Dry Wet Wet Observed trends Climate projections Uncertain Uncertain What is the role Climate change influence on Past event of climate change Drv Wet Dry Wet attribute studies weather patterns role in extreme Uncertain Uncertain precipitation? Climate change influences on Drv Wet thermodynmics

Multiple different potential interpretations

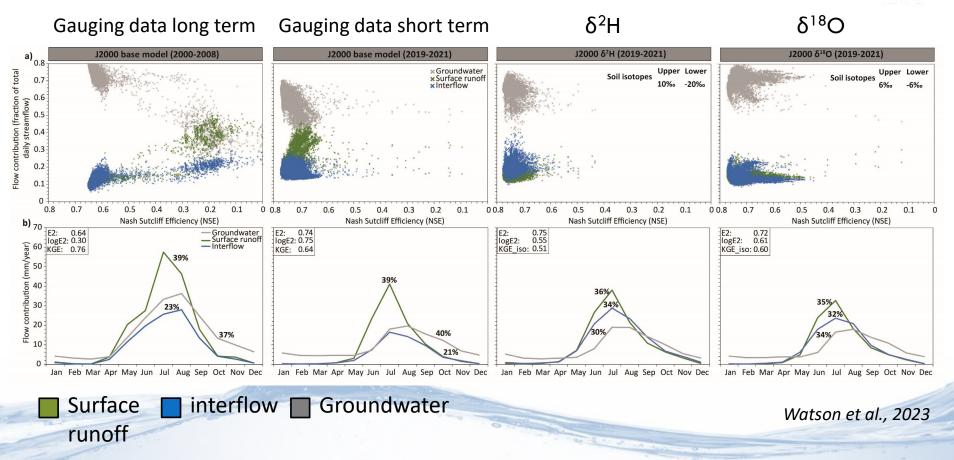
Dr∖

Wet

King et al., 2023, Nature Geoscience

Case Study 1: South Africa



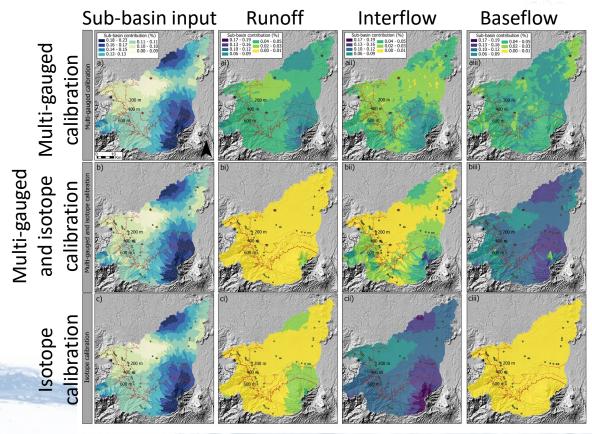


Case Study 2: Costa Rica



Watson et al., submitted

Objective: To evaluate changes in precipitation on surface runoff, interflow and groundwater flow and the ability of isotopes to capture these differences when gauging data were not available. **Isotopes Used:** ²H and ¹⁸O



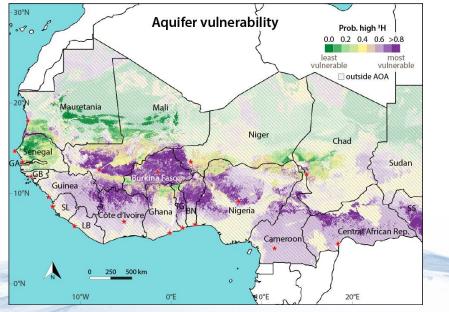
Case Study 2: Tritium

IAEA

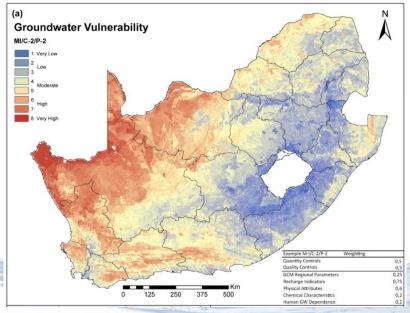
Objective: To understand the risk of groundwater vulnerability to climate change and over-abstraction

Isotope Used: Tritium – ³H

Sahel: Podgorski et al., submitted 2023



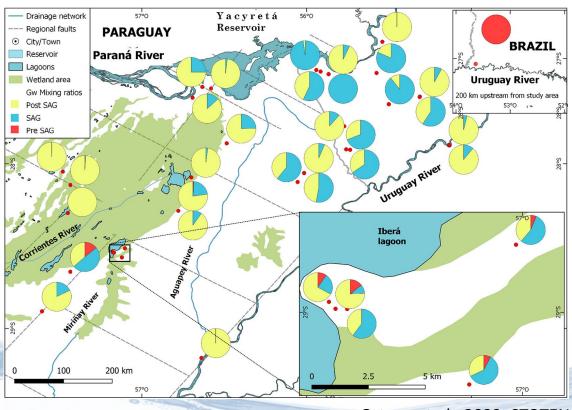
South Africa: van Rooyen et al., 2020



IAEA

Objective: To confirm if discharge from the transboundary Guarani Aquifer System is sustaining the surface water system

Isotope Used: ²²²Radon



Ortega et al., 2023, STOTEN

Isotopic Take Aways

- Simple and inexpensive tools
- Effective at identifying different source regions and mixing relationships
- Can be used to separate atmospheric, surface water and groundwater components
- Can be customized to the situation
- IAEA provides support to Member States for adoption of isotope hydrology approaches







